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# ABSTRACT

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Background: Cardiospermum halicacabum (C. halicacabum) is a common medicinal herb found in India and other Asian countries. It has various medicinal properties such as antimicrobial, pain relief, antibiotics, anti-inflammatory, antioxidants, anticancer etc. It is commonly used for treating diabetes, arthritis, limbs stiffness, rheumatism, lumbago, earache, fever. Type of solvent and polarity intensively affects the antioxidant activity of the extracts due to the solubility of the phytocompounds such as polyphenols and flavonoids in various solvents. Materials and Methods: In this study, different solvents like, ethanol, methanol, chloroform and petroleum ether were used for the extraction of C. halicacabum. Cold maceration method was followed for extraction. The crude extracts were screened preliminary and then confirmed using Fourier transform-infrared spectroscopy analysis. Gas chromatography-mass spectrometry was used to determine the chemical composition of each extract. The DPPH (2.2-diphenyl- 1-picrylhydrazyl) method was used for the evaluation of the antioxidant activity of different crude extracts of C. halicacabum. Results: The results showed that there is significant influence of solvent type in preserving various phytocompounds of the C. halicacabum leaves extract. The evaluation of the antioxidant capacity of different crude extracts was in the order of ethanol > methanol > petroleum ether > chloroform extract.

**Key words**: *Cardiospermum halicacabum*, Medicinal plants, GC-MS, Phytochemicals, Cold maceration, DPPH, Radical scavenging activity.

## INTRODUCTION

Traditionally plants have been used by humans for treating various ailments and is still being used directly or indirectly as supplements.1 Until the development of synthetic drugs, plants have been used for treating almost all diseases. Various studies have reported plants to have medicinal properties such as antimicrobial, pain relief, antibiotics, anti-inflammatory, antioxidants, anticancer etc. Oxidative stress is one among the major factors involved in the rise of degenerative diseases. Recently, interest has increased substantially in finding natural antioxidants for medicinal values and to substitute synthetic antioxidants, which has its limitations of reported carcinogenic effects.<sup>2-5</sup> Oxidative stress and generation of reactive oxygen species (ROS) damages the cell and has been identified to cause extensive damage in tissues leading to various chronic diseases, various degenerative disorders, ageing, cardiovascular diseases, diabetes, Alzheimer's disease, cancer etc.6-11 Oxidative stress also has the ability to cause undesirable effects on food, making it unfit for consumption. Phenolic acid and flavonoids are secondary metabolites of plants which have been excessively studied for their antioxidant and anticancer activities. Various studies show that antioxidants can decrease the risk of various cardiovascular diseases, cancer and arthritis by preventing the oxidative stress.12 Antioxidants scavenge the free radicals present in the biologicals

systems by various reactions such as chelation, quenching the singlet oxygen molecules etc.<sup>13</sup> The presence of antioxidant property in plants has an important role in decreasing the lipid oxidation by scavenging free radicals and by donation of electrons.<sup>14</sup> Apart from antioxidant potential, these secondary metabolites have many more beneficial impacts on human health.

Cardiospermum L. is a genus belonging to Sapindaceae family. It comprises about 17 species of herbs, vines distributed in the tropical regions of the earth.<sup>15</sup> Cardiospermum halicacabum (C. halicacabum) is an herbaceous climber from Sapindaceae family. It is widely distributed in the tropics and subtropics of India, Bangladesh, Pakistan, Malacca, Africa and America. It is one such plant which is traditionally used in Ayurvedic and folk medicines.16 This plant is known to possess various biological and pharmacological properties. Various studies have shown the antimicrobial, antioxidant, analgesic, vasodepressant, anti-inflammatory properties of this plant. It is commonly used for treating arthritis, rheumatism, limbs stiffness, diabetes, lumbago, earache, fever.<sup>17,18</sup> It is the bioactive compounds present in the extracts of leaves, seeds, roots of this plant which contributes for the activities.<sup>19,20</sup> Various pharmacological properties of the phytoconstituents of this plant has been studied. Several chemical constituents such as i.e., apigenin, apigenin-7-Oglucuronide, chrysoeriol-7-O-glucuronide, luteolin-7-O-glucuronide rutin, and β-arachidic acid<sup>21</sup> have

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been isolated from this plant. Various fatty acids isolated from seeds of this plant have been found to be potential in curing neurological disorders.<sup>22</sup> The plant derived herbal products such as gel, cream, medical drops, and pills are available commercially and are useful in treating various chronic skin diseases, including psoriasis and other skin problems.23 The flavonoids isolated from chloroform extract of C. halicacabum have been proved to possess anticancer activities with least hematolytic effects.<sup>24</sup> The presence of phenolic compounds which are known to inhibit the pro-oxidant enzymes has effectively inhibited xanthine oxidase in silk worms.<sup>25</sup> The extraction yield and obtaining the antioxidants from plants depends on factors like solvents used and methods of extraction.<sup>26</sup> The polarity of the solvent used has an impact on preserving the antioxidant activity. Depending on the type of the solvent, the extract can be toxic and may cause undesirable effects. Due to these reasons, it is important to select the solvent of extraction based on the need of phytochemicals and their activities. In a previous study, ethanol, methanol, chloroform and petroleum ether extracts of C. halicacabum was evaluated for the antimicrobial activity. Of which the bactericidal activity was more pronounced in ethanolic extracts compared to other extracts.<sup>27</sup> In another study among benzene, chloroform, ethanol, acetone and aqueous extracts, the alcoholic extract of C. halicacabum showed a wider range of antimicrobial activity against most commensal pathogens.<sup>28</sup> Ponmari et al., 2011 conducted experiments on evaluating the effect of drying methods on the antioxidant activity of C. halicacabum plant extracts.<sup>29</sup> From the literature, it is observed that no efforts have been made towards identifying the effect of solvents on the antioxidant activity of C. halicacabum. This study will hopefully explore and reveal the effects of both polar and non-polar solvents on the preservation of chemical compounds as well as antioxidant activities of this plant.

# **MATERIAL AND METHODS**

### Chemicals

All the chemicals used in this study were of analytical grade and are purchased from Sisco Research Laboratories. The DPPH (2,2-diphenyl-1-picrylhydrazyl) was purchased from Sigma, India

## Preparation of plant extracts

*Cardiospermum halicacabum*, fresh plants were collected in the month of November 2019 from Kanavaipatti village, Namakkal, Tamil Nadu, India. The plant samples were identified and authenticated by Prof. P. Jayaraman, Plant Anatomy Research Centre, Chennai (PARC/2020/4249). The leaves were washed with tap water and then with deionized water thrice to remove dirt. The washed leaves were shade dried under room temperature. A coarse powder of the dried leaves was made using a blending machine and used for solvent extraction. The coarse leaf powder of *C. halicacabum* (each 50g) was extracted with 500ml of ethanol by Cold maceration extraction method for 3 days. The same procedure of extraction was followed for other solvents like methanol, chloroform and petroleum ether. The extracts were concentrated under reduced pressure using a rotary evaporator and stored in an air tight container at cold temperature for future usage.

### Percentage yield

The percentage yield (w/w %) was calculated from the product that was obtained after evaporation using the given formula  $1^{30}$ 

$$Percentage \ yield = \frac{Weight \ of \ the \ product \ obtained \ after \ evaporation}{Weight \ of \ the \ powdered \ sample \ taken \ initially} \times 100$$
(1)

# Phytochemical screening

Phytochemical screening was carried out by methods suggested by Matos<sup>31</sup>, using precipitation and coloration to assess the qualitative phytochemical composition like phenolic compounds, flavonoids, alkaloids, saponins, tannins, reducing sugars, steroids and glycosides

that are present in the crude extracts. The reactions of this screening method showed the presence or absence of the phytochemicals present in crude extracts of various solvents.

#### Fourier Transform Infrared Spectroscopy (FTIR)

Further, it was confirmed by identifying the functional groups present in each extract by FTIR analysis using SHIMADZU IRTracer-100. A small amount of each sample was placed on the crystal and analyzed by attenuated total reflection (ATR) mode. The samples were scanned from 4000 to 400 cm<sup>-1</sup> and the peak values were recorded.

#### Gas Chromatography Mass Spectrometry (GC-MS)

GC-MS spectrometer SHIMADZU, QP2010 PLUS was used to determine the presence and identification of volatile bioactive compounds in each extract. The samples were prepared by dissolving each extract in their respective solvents. The ionization energy of 70 eV was used and Helium gas (99.999%) was used as a carrier with a flow rate of 1.00 ml/min. The injection volume of 2µl was used. The GC injector temperature was maintained at 230°C and the MS transfer line temperature was maintained at 280°C. The temperature for ion source was set at 300 °C. Initial oven temperature was set to 40 °C with a hold time of 60 sec. Then, it was raised to 200 °C (at 5 °C/min) with the hold time of 10 min and then to 235 °C (at 5 °C/ min) with the hold time of 10 min and total running time for each extract was 60 minutes. This last increase was to clean the column from any residues. The results were then analyzed for identification of compounds by matching with available inbuilt libraries; NIST and Willey libraries.

### DPPH radical scavenging activity

The effect of solvents in preserving the antioxidant activity of the *C. halicacabum* leaf extracts was estimated by following the DPPH method suggested by Shimada *et al*,  $1992^{32}$  with slight modifications. The crude extracts of *C. halicacabum* were diluted serially with final concentration as 0.2, 0.4, 0.6, 0.8 and 1.0 mg/ml was prepared. DPPH solution (0.1 mM/L) was freshly prepared by dissolving in methanol. 3.0 ml of the extract with different concentrations was mixed with 1 ml of freshly prepared DPPH solution in a covered test tube. The tubes were incubated in a shaker for 30 min in dark at room temperature. The absorbance at 517nm was measured using UV-Vis spectrophotometer. The scavenging activity was estimated by following the formula 2:

$$DPPH \ scavenging \ Activity (\%) = \frac{(A \ control - A \ sample)}{A \ control} \times 100$$
(2)

where A  $_{control}$  is the absorbance of DPPH solution without extract and A  $_{sample}$  is the absorbance of sample with DPPH solution.

The half-maximal inhibitory concentration (IC<sub>50</sub>) was reported as the amount of antioxidant required to decrease the initial DPPH concentration by 50%. IC<sub>50</sub> value was estimated graphically using non-linear regression algorithm by probit analysis.

#### Statistical analysis

The results of the antioxidant activity are presented as the means  $\pm$  standard deviation (SD) of triplicate assays. The data were subjected to one-way ANOVA by Bonferoni's multiple comparison test and the significance of the difference between means was calculated. All the statistical analysis was performed using GraphPad Prism (version 8) software.

# RESULTS

### Percentage yield

The percentage yield (w/w %) of various extracts of *C. halicacabum* was calculated and was found to be: methanol (18.8 %), ethanol (17.4 %), chloroform (11.8 %) and petroleum ether (12.2 %). From the data, it is

found that the methanol gives the maximum yield and the petroleum ether gives the least yield among the four solvents.

# Phytochemical screening

For phytochemical evaluation, four different solvents were used; all the four solvents were of different polarity. The preliminary qualitative analysis of plant extracts showed the presence of various groups of phytochemicals such as alkaloids, flavonoids, tannins and terpenoids, showed in Table 1. All solvents exhibited a decent amount of the secondary metabolites. Further by FTIR analysis (Figure 1), peaks representing various functional groups were obtained by referring the standards<sup>33,34</sup> and is presented in the Table 2. The peaks indicated the presence of phytocompound groups like polyphenols, flavonoids, tannins, saponins, etc. The results obtained from FTIR analysis are significant with the preliminary qualitative analysis.

# GCMS analysis

The results pertaining to GC-MS analysis lead to the identification of a number of chemical constituents from the GC fractions of solvents extracts of *C. halicacabum*. Twenty-seven compounds from ethanol, 47 from methanol, 69 from chloroform, 83 from petroleum ether were identified based on the retention time. Retention time (RT), % of peak area, molecular formula, molecular weight and names of the unique compounds extracted in each solvent are presented in Table 3.

The compounds with highest area% prevailing in ethanol extract (Figure 2) were Cyclopentanamine (30.06%), Glycerin (22.46%), Hexadecanoic Acid Ethyl Ester (7.63%), 2-Hexadecen-1-Ol, 3,7,11,15-Tetramethyl (5.87%), Neophytadiene (5.74%). The major compounds prevailing in methanol extract (Figure 3) were 9,12,15-Octadecatrienoic Acid Methyl Ester (13.82%), 2-Hexadecen-1-Ol, 3,7,11,15-Tetramethyl- (13.01%), Neophytadiene (8.71%), Hexadecanoic Acid Methyl Ester (6.99%), Cyclopropanebutanoic Acid (4.99%).

The major compounds present in chloroform extract (Figure 4) were Neophytadiene (9.22%), 9-Octadecenamide (7.73), 2-Methyloctacosane (7.21%), 2,4-Ditert-Butylphenol (5.09%). And the major compounds observed in petroleum ether extract (Figure 5) were 9,12,15-Octadecatrienoic Acid (17.66%), N-Hexadecanoic Acid (7.92%), Heneicosane (7.49%), Eicosane (6.16%).

## DPPH radical scavenging activity

The effect of solvent used for extraction on the antioxidant activity was studied by DPPH method. DPPH is widely used for evaluating the antioxidant activities of several natural products in organic solvents within a short time. The percentage of DPPH radical scavenging activity of each extract is presented in Table 4. In this study, the radical scavenging activity (RSA) was found to be concentration dependent. In all the extracts of C. halicacabum, the radical scavenging activity was found to be significantly (p < 0.05, 0.01) more at higher concentration of extracts and less at lower concentration as depicted in Figure 6. The obtained results (Figure 7) showed the levels of significance (p < 0.05) for the effect of solvent on the antioxidant activity of the extracts. No significant difference between ethanol and methanol extracts and no significant difference between chloroform and petroleum ether extracts was reported. Overall, ethanolic extract showed high antioxidant activity followed by methanolic extract, which showed significantly (*p* < 0.05) higher antioxidant activity than chloroform and petroleum ether. The standard ascorbic acid used as the control exhibited maximum inhibition activity of 92.

# *IC*50

The IC<sub>50</sub> is a measure of the effectiveness of a compound in suppressing a particular function. The IC<sub>50</sub> of the DPPH radical scavenging activity of the *C. halicacabum* crude extracts was calculated using well accepted formula and the values are presented in the Table 5. The strongest scavenging activity (244.34  $\mu$ g/ml) was found in the ethanolic crude extracts of *C. halicacabum*.

#### Table 1: Qualitative Phytochemical Analysis of C. halicacabum extracts.

Test -	Positive/Negative				
Test	Ethanol	Methanol	Chloroform	Petroleum Ether	
Reducing Sugar	-	-	-	+	
Glycosides	+	+	+	+	
Tannins	+	+	+	+	
Flavonoids	+	+	-	-	
Terpenoids	-	-	-	+	
Phenols	+	+	+	+	
Saponins	+	+	+	+	

 Table 2: FTIR frequency range and its corresponding functional groups in

 *C. halicacabum* extracts.

S. No	Frequency Ranges	Functional groups
1	3336	O-H stretching Alcohol
2	2924	C-H stretching alkane
3	2856	N-H stretching amine salt
4	2357	Phenyl or amino
5	1712	C-H bending aromatic compound
6	1652	C=C stretching alkene
7	1610	C=C stretching cyclic alkene
8	1452	C-H bending alkane Methyl group
9	1035	S=O stretching sulfoxide
10	840	C=C bending alkene
11	833	1,4-disubstituted or C-H bending

S. No.	Name of the Compound	Retention Time	Peak Area (%)	Molecular Formula	Molecular Weight
Ethanol 1	<i>Extract</i> Cyclopentanamine	7.603	30.06	СНИ	85
2	Glycerin	10.64	22.46	$C_5H_{11}N$ $C_3H_8O_3$	83 92
3	Hexadecanoic Acid, Ethyl Ester	25.417	7.63		284
4	2-Hexadecen-1-Ol, 3,7,11,15-Tetramethyl	27.093	5.87	$C_{18}H_{36}O_2$ $C_{20}H_{40}O$	284
	Neophytadiene				
5		23.185	5.74	$C_{20}H_{38}$	278
6 7	Squalene	38.502	3.4	$C_{30}H_{50}$	410
7	9,12,15-Octadecatrienoic Acid	27.752	2.98	$C_{18}H_{30}O_{2}$	278
8	Hexacosane	42.595	2.48	C <sub>26</sub> H <sub>54</sub>	366
9	2,3-Dihydro-Benzofuran	13.586	2.12	C <sub>8</sub> H <sub>8</sub> O	120
10	3,7,11,15-Tetramethyl-2-Hexadecen-1-Ol	23.826	2.03	C <sub>20</sub> H <sub>40</sub> O	296
11	1,4,5,6-Tetrahydropyrimidine	3.484	1.5	$C_4H_8N_2$	84
12	Pentadecanoic Acid, 14-Methyl-, Methyl Ester	24.488	1.39	$C_{17}H_{34}O_{2}$	270
13	9-Octadecenoic Acid, Methyl Ester	26.891	1.33	$C_{19}H_{36}O_{2}$	296
14	1,2-Dihydro-8-Hydroxylinalool	23.051	1.24	$C_{10}H_{20}O_{2}$	172
15	9,12-Octadecadienoic Acid, Methyl Ester	26.81	1.12	$C_{19}H_{34}O_{2}$	294
16	Octadecanoic Acid, Ethyl Ester	28.062	1.07	$C_{20}H_{40}O_{2}$	312
17	9,12-Octadecadienoic Acid	27.655	1	$C_{18}H_{32}O_{2}$	280
Methano	l Extract				
1	9,12,15-Octadecatrienoic Acid, Methyl Ester	26.932	13.82	$C_{19}H_{32}O_{2}$	292
2	Cyclopropanebutanoic Acid	37.369	4.99	$C_{25}H_{42}O_{2}$	374
3	11,14-Eicosadienoic Acid, Methyl Ester	26.836	4.51	$C_{21}H_{38}O_2$	322
4	9-Octadecenoic Acid	25.564	4.49	$C_{18}H_{34}O_{2}$	282
5	1,2-Benzenedicarboxylic Acid, Dibutyl Ester	25.096	2.96	$C_{16}H_{22}O_{4}$	278
6	Heneicosane	31.274	2.49	C <sub>21</sub> H <sub>44</sub>	296
7	Nonadecane	41.065	2.24	C119H40	268
8	14 BetaH-Pregna	39.326	2.04	$C_{21}H_{36}$	288
9	1,2-Benzenedicarboxylic Acid	34.107	1.97	$C_{24}H_{38}O_{4}$	390
10	2-Pentadecanone, 6,10,14-Trimethyl	23.325	1.82	C <sub>18</sub> H <sub>36</sub> O	268
11	4,8,12,16-Tetramethylheptadecan-4-Olide	30.649	1.43	$C_{21}H_{40}O_{2}$	324
12	3,7,11,15-Tetramethyl-2-Hexadecen-1-Ol	23.559	1.42	$C_{20}H_{40}O$	296
13	(9z,12z,15z)-2-Hydroxy-3-Methoxypropyl Octadeca-9,12,15-Trienoate	19.567	1.25	$C_{22}H_{38}O_{40}$	366
14	2-[(Triethylsilyl)Oxy] Ethanol	7.71	1.22	$C_8 H_{20} O_2 Si$	176
15	2-Tridecenal	23.064	1.11	$C_{13}H_{24}O$	196
16	Tetratetracontane	29.58	1.07	$C_{44}H_{90}$	618
17	2,3-Dihydro-Benzofuran	13.63	1.05	$C_{8}H_{8}O$	120
	rm Extract	10.00	1.00	0,811,80	120
1	9-Octadecenamide	38.335	7.73	C <sub>18</sub> H <sub>35</sub> NO	281
2	2-Methyloctacosane	42.594	7.21	$C_{29}H_{60}$	408
3	2,4-Ditert-Butylphenol	18.334	5.09	$C_{14}H_{29}O$	206
4	E-14-Hexadecenal	22.532	3.28	$C_{14} H_{22} C_{16} H_{30} O$	238
5	1-Nonadecene	28.07	2.41	$C_{16}H_{30}$	266
6	1-Hexadecene	19.34	2.38	$C_{19}T_{38}$ $C_{16}H_{32}$	224
7	Pentacosane	29.568	2.33	$C_{16}H_{32}$ $C_{25}H_{52}$	352
8	Octacosane	36.846	2.35	$C_{25}^{-1}C_{28}^{-1}H_{58}^{-1}$	394
9	E-15-Heptadecenal	25.42	2.2	$C_{28}H_{58}$ $C_{17}H_{32}O$	252
	Tritetracontane				
10		41.045	1.57	$C_{43}H_{88}$	604 256
11	Tetradecanoic Acid, 12-Methyl-, Methyl Ester	24.493	1.54	$C_{16}H_{32}O_{2}$	256
12	Octadecanamide	31.443	1.49	C <sub>18</sub> H <sub>37</sub> NO	283
13	1-Tetradecene	15.785	1.29	$C_{14}H_{28}$	196
	m Ether Extract	20.200	15.44		250
1	9,12,15-Octadecatrienoic Acid	28.398	17.66	$C_{18}H_{30}O_{2}$	278
2	N-Hexadecanoic Acid	25.886	7.92	$C_{16}H_{32}O_{2}$	256
3	Heneicosane	42.825	7.49	C <sub>21</sub> H <sub>44</sub>	296
4	Eicosane	33.463	6.16	$C_{20}H_{42}$	282
5	2-Hexadecen-1-Ol, 3,7,11,15-Tetramethyl	27.213	4.22	$C_{20}H_{40}O$	296
6	2-Pentadecanone, 6,10,14-Trimethyl	23.339	1.44	$C_{18}H_{36}O$	268
7	Bicyclo[7.2.0]Undec-4-Ene, 4,11,11-Trimethyl-8-Methylene	16.579	1.35	$C_{15}H_{24}$	204
8	Tetradecanamide	31.66	1.26	$C_{14}H_{29}NO$	227
9	Hexadecanoic Acid, Methyl Ester	24.516	1.13	C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>	270

# Table 3: GC-MS Profile of Different Solvent Extracts of C. halicacabum.

Solvent -	Antioxidant Activity (%)					
	200 µg/ml	400 µg/ml	600 µg/ml	800 μg/ml	1000 µg/ml	
Ascorbic acid	$53.86 \pm 1.23$	$64.11 \pm 1.05$	$73.92 \pm 1.32$	$85.65 \pm 1.12$	92.13 ± 1.28	
Ethanol	$48.24\pm0.23$	$59.62 \pm 1.02$	$63.96 \pm 0.94$	$69.92 \pm 1.41$	$79.00 \pm 1.02$	
Methanol	$47.56 \pm 1.08$	$55.56\pm0.85$	$61.52\pm0.23$	$65.72 \pm 1.24$	$76.42 \pm 1.08$	
Chloroform	$22.36\pm0.70$	$28.46 \pm 0.70$	$37.40 \pm 1.08$	$52.57 \pm 1.43$	$55.28 \pm 0.41$	
Petroleum Ether	$26.42\pm0.70$	$37.13 \pm 0.23$	$49.05\pm0.62$	$53.52 \pm 1.02$	$56.37 \pm 1.24$	

#### Table 4: DPPH scavenging activity.

Note: The values are the mean ± standard deviation for triplicate experiments.

# Table 5: IC<sub>50</sub> in DPPH radical scavenging activity.

Extracts	IC <sub>50</sub> (μg/ml)
Ethanol	244.34
Methanol	272.20
Chloroform	1141.03
Petroleum ether	816.01

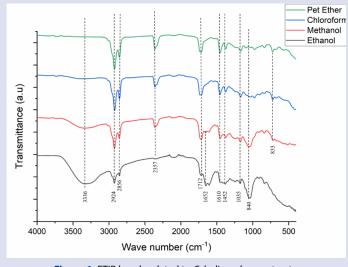
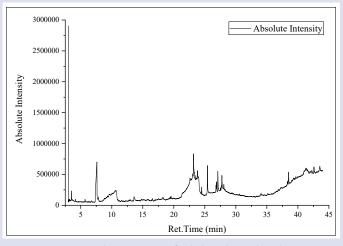
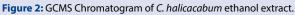


Figure 1: FTIR bands related to C. halicacabum extracts.





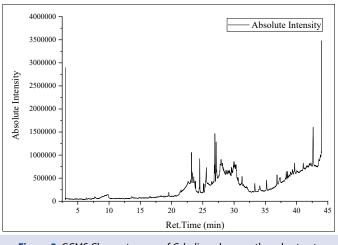


Figure 3: GCMS Chromatogram of C. halicacabum methanol extract.

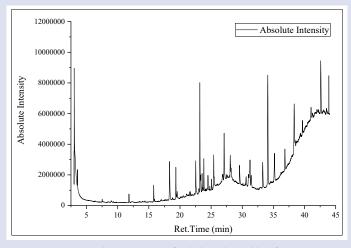
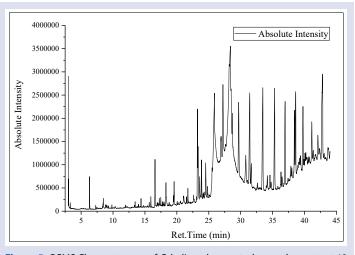
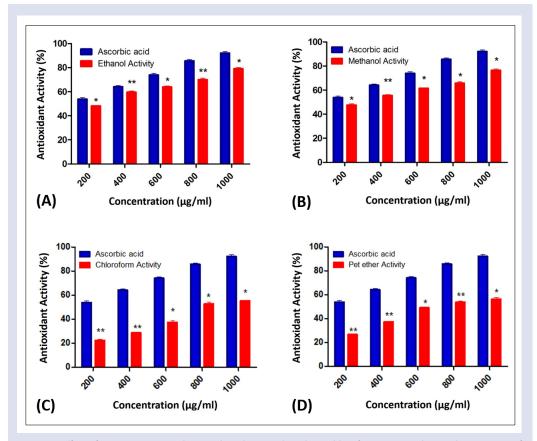
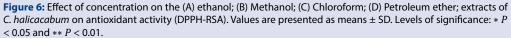


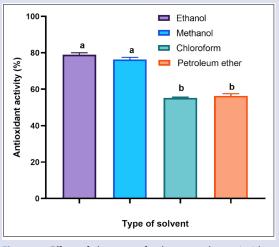
Figure 4: GCMS Chromatogram of C. halicacabum chloroform extract.

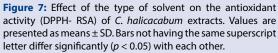


**Figure 5:** GCMS Chromatogram of *C. halicacabum* petroleum ether extract 13  $\pm$  1.28 at higher concentration.









# DISCUSSION

Phenols and flavonoids are the representative groups for the antioxidant compounds. The presence of phenols and flavonoids revealed by preliminary phytochemical evaluation and confirmed by the FTIR analysis supports the presence of antioxidant compounds in the C. halicacabum extracts. These results are found to be in accordance with Viji et al., 2010. Further, the GCMS analysis of C. halicacabum leaves extracts showed the presence of different phytochemicals in each solvent extracts. Many studies have used GCMS analysis to show the impact of solvents on the isolation of phytocompounds with medicinal importance from the crude plant extracts. Our GCMS results are in accordance with that of previously reported.35-38 These compounds identified in our study belong to tannins, saponins, glycosides, fatty acids, esters of fatty acids, etc. Various therapeutic properties of the identified compounds have been reported and are being used in medical formulation and various other applications. For example, the compounds such as 9,12-Octadecadienoic Acid, Methyl Ester (Linolenic acid), n-Hexadecanoic acid (Palmitic acid), Hexadecanoic Acid Ethyl Ester, Hexadecanoic Acid Methyl Ester, Tetradecanoic Acid, 12-Methyl-, Methyl Ester, squalene are known to possess antioxidant and 5-Alpha reductase inhibitor activity.<sup>39,40</sup> Compounds like 3,7,11,15-tetramethyl-2-hexadecen-1-ol (phytol), squalene also has antimicrobial activities.<sup>41,42</sup> Compounds like esters of fatty acids with simple OH group are of commercial importance and are used as texturizing, emulsifying, antifoam and stabilizing agents and also has its applications in cosmetics, detergents and pharma products due to its fragrance.<sup>43,44</sup> Since GCMS analysis shows only the presence of volatile compounds, phenolic compounds being non-volatile did not show up in the results of GCMS analysis.

Based on the solvent used for extraction, variations in radical scavenging ability of the *C. halicacabum* extracts was found. This shows that there is a significant influence of the solvent on antioxidant activity of *C. halicacabum* crude extracts. These findings were supported by previous studies on *Macademia tetraphylla* peels, selected herbal infusions, *Limnophila aromatic* and *Withnaia somnifera* (L.), which reported that there is a strong influence of solvent on the antioxidant activity of plant extracts.<sup>45-48</sup> This variation of antioxidant activity can be possibly associated with the plant compounds like phenolic constituents which are soluble based on the polarity of the solvents used for extraction.<sup>49-50</sup> Thus, making the extracts obtained from polar solvent more vital in antioxidant activity.<sup>51</sup> Because of the presence of strong radical scavenging activity, alcoholic extracts of *C. halicacabum* is a potential source of natural antioxidants and it can be used to ameliorate the effects of oxidative stress caused by ROS.

### CONCLUSION

Studies based on the bioactive components from plants and their activities are increasing nowadays. The GCMS data revealed the presence of various bioactive compounds. These bioactive compounds are known to have noble biological activities. This study showed the presence of antioxidant activities of the crude extracts of *C. halicacabum*. And the IC<sub>50</sub> values were also comparatively significant. The extract obtained from polar solvents like ethanol and methanol showed highest antioxidant activity. This clearly indicates that there is a significant effect of the solvent used for extraction. Ethanolic extract of *C. halicacabum* is a most potential extract and is a promising source of antioxidant agents. Further work regarding the identification of non-volatile compounds and their specific activity will give us insight about the usage.

### **CONFLICTS OF INTEREST**

The authors declare that they have no conflicts of interest.

### ABBREVIATIONS

GCMS: Gas chromatography-mass spectrometry; FTIR: Fourier transform infrared; ATR: Attenuated total reflectance; DPPH: 2,2-diphenyl- 1-picrylhydrazyl; UV-Vis: Ultra-violet Visible; NIST: National Institute of Standards and Technology; ANOVA: Analysis of variation; RSA: Radical scavenging activity.

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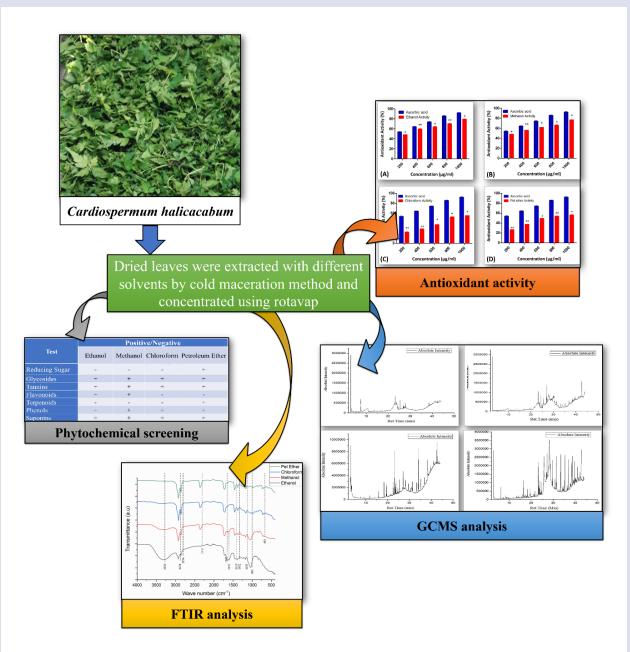
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# **GRAPHICAL ABSTRACT**



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